## Math 111: Final Review 5b

1. In chemistry, C denotes the concentration of hydrogen ions (measured in moles/liter) in a solution;  $C \in (0, 1)$ . The pH of a solution is calculated as:

$$pH = -\log C$$

a) A Rockstar energy drink is acidic and has a pH = 2.53. Find the concentration of hydrogen ions in a solution with a pH of 2.53.

**b)** Fiji water is a slightly alkaline solution and has a pH = 7.5. Find the concentration of hydrogen ions in a solution with a pH of 7.5.

c) Find the inverse function for the function pH(C) including the domain & range.

d) How many times greater is the hydrogen ion concentration in a Monster energy drink with a pH = 3.49 than the hydrogen ion concentration in mild with a pH = 7.33? State the exact answer, then give the answer rounded to two places.

$$3.49 = -\log C$$

$$C = 10^{3.49}$$

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$$C = 10^{3.49}$$

$$C = 10^{3.84}$$

2. In seismology, I is the intensity measurement of the wave energy of an earthquake; I > 0. The *Richter Scale measurement* of an earthquake can be approximated with:

$$R = \log I$$

**a)** The 1906 San Francisco earthquake had a Richter scale measure of 8.6. Find its intensity.

**b)** In November, 2016 there was an earthquake in Warrenton, OR that had a Richter scale measure of 3.6. Find its intensity.

c) Find the inverse function for the function R(I) including the domain & range.

**d)** How many times greater is the intensity of the San Francisco Earthquake than the Warrenton one? State the exact answer, then give the answer rounded to two places.

$$\frac{10^{8.6}}{10^{3.6}}$$
 = 100,000 = 10<sup>5</sup> times greater